Week 1	Weekly learning outcomes	Student book links	Practical activity links
 Revision of forces and acceleration Newton's first law Momentum Newton's second law Impulse Newton's third law 	 Students should be able to: State and use each of Newton's three laws of motion. Define <i>linear momentum</i> as the product of mass and velocity (<i>ρ</i> = <i>mv</i>). Define <i>net force on a body</i> as equal to the rate of change of its momentum. Select and apply the equation <i>F</i> = Δ<i>p</i>/Δ<i>t</i> to solve problems. Explain that <i>F</i> = <i>ma</i> is a special case of Newton's second law when mass <i>m</i> remains constant. Define <i>impulse of a force</i>. Recall that the area under a force against time graph is equal to impulse. Recall and use the equation <i>impulse</i> = <i>change in momentum</i>. 	G484 The Newto 4.1.1 Newton's law	Work topic outlines nian World ws of motion – Newton's three laws of ar momentum, and impulse as a force

leek 2	Weekly learning outcomes	Student book links	Practical activity links
 Conservation of momentum Examples and applications of conservation of momentum Elastic and inelastic collisions 	 Students should be able to: State the principle of conservation of momentum. Apply the principle of conservation of momentum to solve problems when bodies interact in one dimension. Define a perfectly <i>elastic collision</i> and an <i>inelastic collision</i>. Explain that whilst the momentum of a system is always conserved in the interaction between bodies, some change in kinetic energy usually occurs. 	 1.1.5 1.1.7 OCR Scheme of G484 The Newto 4.1.2 Conservation 	Practical activity 1: Collisions on a linear air track Practical activity 2: The initial velocity of an air gun pellet Practical activity 3: The bounce height of a ball



Week 3	Weekly learning outcomes	Student book links	Practical activity links
 The radian Motion in a circle Centripetal acceleration Centripetal force Examples of circular motion 	 Students should be able to: Define the <i>radian</i>. Convert angles from degrees into radians and vice versa. Explain that a force perpendicular to the velocity of an object will make the object describe a circular path. Explain what is meant by centripetal acceleration and centripetal force. Select and apply the equations for speed and centripetal acceleration: v = 2πr/T and a = v²/r. Select and apply the equation for centripetal force: F = ma = mv²/r. 	G484 The Newto 4.2.1 Circular mot and degrees	Practical activity 4: Motion in a circle Work topic outlines mian World tion – Define the radian, use radians , circular motion, centripetal centripetal force.

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Week 4	Weekly learning outcomes	Student book links	Practical activity links
 Revision of free fall Gravitational fields Newton's law of gravitation Gravitational field strength The Earth's gravitational field and gravitational field strength 	 Students should be able to: Describe how a mass creates a gravitational field in the space around it. Define <i>gravitational field strength</i> as force per unit mass. Use gravitational field lines to represent a gravitational field. State Newton's law of gravitation and select and use the equation for the force between two points or spherical objects: <i>F</i> = -<i>GMm/r</i>². Select and apply the equation for the gravitational field strength of a point mass: <i>g</i> = -<i>GM/r</i>². Select and use the equation to determine the mass of the Earth or another similar object: <i>g</i> = -<i>GM/r</i>². Explain that close to the Earth's surface, the gravitational field strength is uniform and approximately equal to the acceleration of free fall. 	G484 The Newton	l fields – Gravitational fields, Newton's

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,	Veek 5	Weekly learning outcomes	Student book links	Practical activity links
	 Planetary orbits – simple, circular motion, etc. Kepler's third law Planetary orbits – Kepler's third law and Newton's law of gravitation Satellites 	 Students should be able to: Analyse circular orbits in an inverse square law field by relating the gravitational force to the centripetal acceleration it causes. Define and use the <i>period of an object describing a circle</i>. Derive the equation T² = (4π²/GM)r³ from first principles. Select and apply the equation above for planets and satellites – natural and artificial. Select and apply Kepler's third law to solve problems. Define the <i>geostationary orbit</i> of a satellite and state the uses of such satellites. 	 1.2.8–9 OCR Scheme of V G484 The Newton 4.2.2 Gravitational 	Work topic outlines

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Week 6	Weekly learning outcomes	Student book links	Practical activity links
 Free oscillations Simple harmonic motion (SHM) – definition and simple examples SHM – equations SHM – graphs 	 Students should be able to: Describe simple examples of free oscillations. Define and use the terms <i>displacement</i>; <i>amplitude</i>; <i>period</i>; <i>frequency</i>; <i>angular</i> <i>frequency</i>; and <i>phase difference</i>. Select and use the equation <i>period</i> = 1/<i>frequency</i>. Define <i>simple harmonic motion</i> (SHM). Select and apply the equation a = - (2πl)²x as the defining equation of SHM. Select and use x = Acos(2πft) or x = Asin(2πft) as solutions to the equation a = - (2πl)²x. Select and apply the equation v_{max} = (2πl)A for the maximum speed of a simple harmonic oscillator. Explain that the period of an object with SHM is independent of its amplitude. Describe with graphical illustrations the changes in displacement, velocity and acceleration during SHM. 	G484 The Newtor	onic oscillations – Free oscillation,

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Week 7	Weekly learning outcomes	Student book links	Practical activity links
 Energy in simple harmonic motion (SHM) Damping Forced oscillations Resonance Applications and problems of resonance 	 Students should be able to: Describe and explain the interchange between kinetic and potential energy during simple harmonic motion (SHM). Describe the effects of damping on an oscillatory system. Describe practical examples of forced oscillations and resonance. Describe graphically how the amplitude of a forced oscillation changes with frequency near to the natural frequency of the system. Describe examples where resonance is useful and other examples where resonance should be avoided. 	G484 The Newtor 4.2.3 Simple harm simple harm	Practical activity 9: Simple damping Practical activity 10: The damping of a light-beam galvanometer movement Practical activity 11: Barton's pendulums Practical activity 12: The forced oscillations of a mass spring Practical activity 13: The forced oscillations of a lightly damped light- beam galvanometer movement Work topic outlines hian World onic oscillations – Energy changes in onic motion, damping on an ystem, and resonance

Week 8	Weekly learning outcomes	Student book links	Practical activity links
 Density and states of matter (particles) Brownian motion Kinetic theory of matter Internal energy Changes of state 	 Students should be able to: Describe solids, liquids and gases in terms of the spacing, ordering and motion of atoms or molecules. Describe a simple kinetic model for solids, liquids and gases. Describe an experiment that demonstrates Brownian motion and discuss the evidence for the movement of molecules provided by such an experiment. Define the term <i>pressure</i> and use the kinetic model to explain the pressure exerted by gases. Define <i>internal energy</i> as the sum of the random distribution of kinetic and potential energies associated with the molecules of a system. Explain that the rise in temperature of a body leads to an increase in its internal energy. Explain that a change of state for a substance leads to changes in its internal energy but not its temperature. Describe, using a simple kinetic model for matter, the terms: <i>melting</i>; <i>boiling</i>; and <i>evaporation</i>. 	G484 The Newton 4.3.1 Solid, liquid motion, simp	Work topic outlines hian World and gas – Spacing and ordering, ble kinetic model, pressure, internal changes of state

Week 9	Weekly learning outcomes	Student book links	Practical activity links
 Heat and temperature Absolute zero and Kelvin Specific heat capacity – definition and examples Measuring specific heat capacity Latent heat of fusion and vaporisation 	 Students should be able to: Explain that thermal energy is transferred from a region of higher temperature to a region of lower temperature. Explain that regions of equal temperature are in thermal equilibrium. Describe how there is an absolute scale of temperature that does not depend on the property of any particular substance – i.e. the thermodynamic scale and the concept of absolute zero. Convert temperatures measured in kelvins to degrees Celsius (or vice versa): T (K)= θ (°C) + 273.15 State that absolute zero is the temperature at which a substance has minimum internal energy. Define and apply the concept of <i>specific heat capacity</i>. Select and apply the equation <i>E</i> = mcΔθ. Describe an electrical experiment to determine the specific heat capacity of a solid or a liquid. Describe what is meant by the terms: <i>latent heat of fusion</i> and <i>latent heat of vaporisation</i>. 	G484 The Newtor 4.3.2 Temperature degrees Cels 4.3.3 Thermal prop	Practical activity 15: An electrical method to measure a specific heat capacity Practical activity 16: Determining specific heat capacity by other methods Mork topic outlines han World - Thermal equilibrium, Kelvin and sius, and absolute zero perties of materials – Specific heat latent heat of fusion and vaporisation

W	eek 10	Weekly learning outcomes	Student book links	Practical activity links
 1. 2. 3. 4. 5. 6. 		 Weekly learning outcomes Students should be able to: State Boyle's law. Select and apply <i>pV/T</i> = constant. State the basic assumptions of the kinetic theory of gases. State that one mole of any substance contains 6.02 × 10²³ particles and that 6.02 × 10²³ mol⁻¹ is the Avogadro constant <i>N_A</i>. Select and solve problems using the ideal gas equation expressed as: <i>pV</i> = <i>NkT</i> and <i>pV</i> = <i>nRT</i>, where <i>N</i> is the number of atoms and <i>n</i> is the number of moles. Explain that the mean translational kinetic energy of an atom of an ideal gas is directly proportional to the temperature of the gas in kelvins. Select and apply the equation <i>E</i> = 3/2 <i>kT</i>, the mean translational kinetic energy of atoms. 	links • 1.3.8–11 OCR Scheme of V G484 The Newton 4.3.4 Boyle's law -	Practical activity 14: Gas law experiments Work topic outlines

Week 11	Weekly learning outcomes	Student book links	Practical activity links
 Electric fields – definition Electric fields – shapes and properties Coulomb's law Electric field strength around point charge Uniform electric fields Charged particles in electric fields Comparison between gravitational fields and electric fields 	 Students should be able to: State that electric fields are created by electric charges. Define <i>electric field strength</i> as force per unit positive charge. Describe how electric field lines represent an electric field. Select and use Coulomb's law in the form <i>F</i> = <i>Qq</i>/4<i>πε</i>₀<i>r</i>². Select and apply <i>E</i> = <i>Q</i>/4<i>πε</i>₀<i>r</i>² for the electric field strength of a point charge. Select and use <i>E</i> = <i>V</i>/<i>d</i> for the magnitude of the uniform electric field strength between charged parallel plates. Explain the effect of a uniform electric field on the motion of charged particles. Describe the similarities and differences between the gravitational fields of point masses and the electric fields of point charges. 	G485 Fields, Part 5.1.1 Electric fields electric fields electric fields	Practical activity 17: Electric field patterns Practical activity 18: Measuring the constant in Coulomb's law Work topic outlines Eicles and Frontiers of Physics s – Electric fields, electric charge, strength, Coulomb's law, uniform and the motion of charged particles, gravitational and electric fields

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Week 12	Weekly learning outcomes	Student book links	Practical activity links	
 Magnetic fields – shapes and properties Magnetic field around a current-carrying wire Electromagnets Forces on current-carrying wires in magnetic fields (FLHR and <i>F</i> = <i>BIL</i>) 	 Students should be able to: Describe the magnetic field patterns of a long, straight, current-carrying conductor and a long solenoid. State and use Fleming's left-hand rule to determine the force on a current conductor placed at right angles to a magnetic field. Select and use the equations <i>F</i> = <i>BIL</i> and <i>F</i> = <i>BILsinθ</i>. Define <i>magnetic flux density</i> and the <i>tesla</i>. 	• 2.1.4–6	Practical activity 19: Investigating magnetic fields around conductors carrying a current Practical activity 20: The force on a wire carrying a current in a magnetic field	
5. Magnetic flux density		OCR Scheme of Work topic outlines		
		G485 Fields, Particles and Frontiers of Physics		
		a long solen	lds – Magnetic field patterns, fields of oid, Fleming's left-hand rule, x density and tesla	

 1. Force on charged particles in magnetic fields 2. Magnetic fields and circular motion 3. Applications of electric and magnetic fields – mass spectrometer Students should be able to: Select and use the equation <i>F</i> = <i>BQv</i> for the force on a charged particle travelling at right angles to a uniform magnetic field. Analyse the circular orbits of charged particles moving in a plane perpendicular to a uniform magnetic force to the centripetal acceleration it causes. Analyse the motion of charged particles in both electric and magnetic fields. Explain the use of deflection of charged particles in the magnetic and electric fields of a mass spectrometer. Category and the spectrometer of the magnetic and electric fields of a mass spectrometer. Category and the spectrometer of the magnetic fields. Explain the use of deflection of charged particles in both electric fields of a mass spectrometer. Category and the magnetic and electric fields of a mass spectrometer. Category and the magnetic and electric fields of a mass spectrometer. Category and the spectrometer of the magnetic fields of a mass spectrometer. Category and the spectrometer of the magnetic fields of a mass spectrometer. 	Week 13	Weekly learning outcomes	Student book links	Practical activity links
	 Force on charged particles in magnetic fields Magnetic fields and circular motion Applications of electric and magnetic fields – mass 	 Students should be able to: Select and use the equation <i>F</i> = <i>BQv</i> for the force on a charged particle travelling at right angles to a uniform magnetic field. Analyse the circular orbits of charged particles moving in a plane perpendicular to a uniform magnetic field by relating the magnetic force to the centripetal acceleration it causes. Analyse the motion of charged particles in both electric and magnetic fields. Explain the use of deflection of charged particles of 	 2.1.7–8 OCR Scheme of V G485 Fields, Part 5.1.2 Magnetic fie particles, ch magnetic fie 	Work topic outlines ticles and Frontiers of Physics lds – Circular orbits of charged arged particles in electric and elds, and deflection of charged

	ks	Practical activity links
Ilectromagnetic induction FRHR)Define magnetic flux.Magnetic flux and the veberDefine the weber.Magnetic flux and the veberSelect and use the equation for magnetic flux: $\phi = BAcos \theta$.Magnetic flux linkage faraday's law of lectromagnetic induction - imple statements and xperimentsDefine magnetic flux.State and use Faraday's law of electromagnetic induction.OCRState and use Faraday's law of electromagnetic induction.G488	2.1.9–11 R Scheme of V 85 Fields, Parti .3 Electromagne magnetic flux	Practical activity 22: Observing induced voltages Vork topic outlines induced rontiers of Physics etism – Magnetic flux and the weber, inkage, Faraday's law of etic induction, Lenz's law, and simple ors

Week 15	Weekly learning outcomes	Student book links	Practical activity links
 Transformers – structure and basic use Operation of transformers Transformer equation Efficiency of transformers 	 Students should be able to: Describe the function of a simple transformer. Select and use the turns–ratio equation for a transformer. Describe the function of step-up and step-down transformers. 	2.1.12–13 Practical activity 23: Investigating electromagnetic induction using changing fields Practical activity 24: Investigating action of a transformer	
		OCR Scheme of	Work topic outlines
		5.1.3 Electromagn magnetic flu electromagn	ticles and Frontiers of Physics netism – Magnetic flux and the weber, x linkage, Faraday's law of netic induction, Lenz's law, s, and simple a.c. generators

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Week 16	Weekly learning outcomes	Student book links	Practical activity links
 The capacitor – basic structure and principles of use Charging capacitors Capacitance – definition and equation Combining capacitors in series Combining capacitors in parallel Energy stored in capacitors Energy stored in capacitor combinations 	 Students should be able to: Define <i>capacitance</i> and the <i>farad</i>. Select and use the equation Q = VC. State and use the equation for the total capacitance of two or more capacitors in series. State and use the equation for the total capacitance of two or more capacitors in parallel. Solve circuit problems with capacitors involving series and parallel circuits. Explain that the area under a potential difference against charge graph is equal to energy stored by a capacitor. Select and use the equations W = ½ QV and W = ½ CV² for a charged capacitor. 	G485 Fields, Part 5.2.1 Capacitors – storage in a c	Practical activity 25: The capacitor equation Practical activity 26: The rules of addition for capacitors in series and parallel Practical activity 27: Energy transfer using a capacitor Work topic outlines ticles and Frontiers of Physics - Capacitance and the farad, energy capacitor, total capacitance, and series and parallel circuits

Week 17	Weekly learning outcomes	Student book links	Practical activity links
 Discharging a capacitor – qualitative description Time constant Exponential decay Use of the exponential equations Uses of capacitors 	 Students should be able to: Define the <i>time constant</i> of a circuit. Select and use <i>time constant = CR</i>. Analyse the discharge of a capacitor using equations of the form x = x_o e^{-t/CR}. Explain exponential decays as having a constant-ratio property. 	• 2.2.3–5	Practical activity 28: The charging and discharging a capacitor through a resistor Practical activity 29: To determine the initial velocity of an air gun pellet using a C–R circuit
	 Describe the uses of capacitors for the storage of energy in applications such as flash photography, lasers used in nuclear fusion and as backup power supplies for computers. 	OCR Scheme of Work topic outlines	
		5.2.1 Capacitors - resistor, tim	ticles and Frontiers of Physics – Discharge of a capacitor through a e constant of a circuit, and uses of or the storage of energy

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Week 18	Weekly learning outcomes	Student book links	Practical activity links
 History of atom models Alpha scattering The nucleus – components and forces Nuclides, elements and 	 Students should be able to: Describe qualitatively the alpha particle scattering experiment and the evidence this provides for the existence, charge and small size of the nucleus. 	• 2.3.1–4	
isotopes 5. Changes to the nucleus	 Describe the basic atomic structure of the atom and the relative sizes of the atom and the nucleus. Select and use Coulomb's law to determine the force of repulsion and Newton's law of gravitation to determine the force of attraction between two protons at nuclear separations and hence the need for a short-range, attractive force between nucleons. Describe how the strong nuclear force between nucleons is attractive and very short-ranged. Estimate the density of nuclear matter. Define <i>proton</i> and <i>nucleon number</i>. State and use the <i>A</i>, <i>Z</i> notation representation of nucleas. Define and use the term <i>isotope</i>. Use nuclear decay equations to represent simple nuclear reactions. 	OCR Scheme of Work topic outlines	
		5.3.1 The nuclear experiment, Coulomb's la strong nucle	ticles and Frontiers of Physics atom – Alpha particle scattering atomic structure of the atom, aw, Newton's law of gravitation, ar force, density of nuclear matter, ber, nucleon number, isotopes, and ay
	 State the quantities conserved in a nuclear decay. 		

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Week 19	Weekly learning outcomes	Student book links	Practical activity links
 Introduction to the particle zoo Quarks Combining quarks – baryons and mesons Antimatter Beta decay in terms of quarks Leptons 	 Students should be able to: Explain that since protons and neutrons contain charged constituents called quarks they are therefore <i>not</i> fundamental particles. Describe a simple quark model of hadrons in terms of up, down and strange quarks and their respective antiquarks – taking into account their charge, baryon number and strangeness. Describe how the quark model may be extended to include the properties of charm, topness and bottomness. Describe the properties of neutrons and protons in terms of a simple quark model. Describe how there is a weak interaction between quarks and that this is responsible for beta decay. State that there are two types of beta decay. Describe the two types of beta (β) decay in terms of a simple quark model. State that (electron) neutrinos and (electron) antineutrinos are produced during β⁺ and β⁻ decays, respectively. State that a β⁻ particle is an electron and a β⁺ particle is a positron. State that electrons and neutrinos are members of a group of particles known as leptons. 	G485 Fields, Part 5.3.2 Fundamenta	Work topic outlines icles and Frontiers of Physics I particles – Protons, neutrons and k model of hadrons, leptons, and β

Week 20	Weekly learning outcomes	Student book links	Practical activity links
 Nature of radioactive decay Activity and the becquerel Decay constant – introduction as probability of decay Alpha particles and alpha decay Beta particles and beta decay Gamma rays and gamma decay Exponential decay 	 Students should be able to: Describe the spontaneous and random nature of radioactive decay of unstable nuclei. Describe the nature, penetration and range of alpha particles, beta particles and gamma rays. Define and use the quantities <i>activity</i> and <i>decay constant</i>. Select and apply the equation for activity A = λN. Select and apply the equations A = A₀e^{-λt} and N = N₀e^{-λt} where A is the activity and N is the number of undecayed nuclei. 	G485 Fields, Part 5.3.3 Radioactivity	Practical activity 30: Experiments with alpha particles Practical activity 31: Experiments with beta particles Practical activity 32: The absorption of gamma rays Work topic outlines icles and Frontiers of Physics - Radioactive decay, α-particles, A-rays, decay constant, and activity

Week 21	Weekly learning outcomes	Student book links	Practical activity links	
 Week 21 Half-life – general definition and determination from activity–time graphs Half-life and the decay constant Uses of radioactivity Radioactive dating Comparison between radioactive decay and capacitor discharge 	 Weekly learning outcomes Students should be able to: Define and apply the term <i>half-life</i>. Select and use the equation <i>\lambda t_{1/2} = 0.693</i>. Compare and contrast the decay of radioactive nuclei and the decay of charge on a capacitor in a C–R circuit. Describe the use of radioactive isotopes in smoke alarms. Describe the technique of radioactive dating – i.e. carbon dating. 	Described and the first set		

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Week 22	Weekly learning outcomes	Student book links	Practical activity links
 E = mc² Missing mass and binding energy Binding energy per nucleon Curve of binding energy 	 Students should be able to: Select and use Einstein's mass–energy equation ΔE = Δmc². Define <i>binding energy</i> and <i>binding energy per nucleon</i>. Use and interpret the binding energy per nucleon against nucleon number graph. Determine the binding energy of nuclei using ΔE = Δmc² and masses of nuclei. 	G485 Fields, Part	Nork topic outlines icles and Frontiers of Physics on and fusion – Einstein's y equation

Week 23	Weekly learning outcomes	Student book links	Practical activity links
 Nuclear fission – basic description and energy released per reaction Chain reaction Nuclear fission reactors Nuclear fusion – basic description and energy released per reaction Nuclear fusion reactors and fusion in stars 	 Students should be able to: Describe the process of induced nuclear fission. Describe and explain the process of nuclear chain reaction. Describe the basic construction of a fission reactor and explain the role of the fuel rods, control rods and the moderator. Describe the use of nuclear fission as an energy source. Describe the peaceful and destructive uses of nuclear fission. Describe the environmental effects of nuclear waste. Describe the process of nuclear fusion. Describe the environmental effects of stars that make fusion possible. Calculate the energy released in simple nuclear reactions. 	G485 Fields, Part 5.4.1 X-rays – X-ra effect, pair p	Work topic outlines ticles and Frontiers of Physics ays, photoelectric effect, Compton roduction, intensity, X-ray imaging, rs, medical tracers, gamma camera,

Week 24	Weekly learning outcomes	Student book links	Practical activity links
 Description of X-rays – revision of photon model Production of X-rays X-ray interactions – Compton effect, pair production and photoelectric effect X-ray intensity Computerised axial tomography (CAT) scans Limitations of X-rays – soft tissue, ionising, etc. – and the alternatives 	 Students should be able to: Describe the nature of X-rays. Describe in simple terms how X-rays are produced. Describe how X-rays interact with matter – limit to photoelectric effect, Compton effect and pair production. Define intensity as the power per unit crosssectional area. Select and use the equation <i>I</i> =<i>I</i>₀e^{-μx} to show how the intensity <i>I</i> of a collimated X-ray beam varies with thickness <i>x</i> of the medium. Describe the use of X-rays in imaging internal body structures including the use of image intensifiers and of contrast media. Explain how soft tissues such as the intestines can be imaged using a barium meal. Describe the advantages of a CAT scan compared with an X-ray image. 	G485 Fields, Par 5.4.1 X-rays – X-r effect, pair p CAT scanne 5.4.2 Diagnosis m	Work topic outlines ticles and Frontiers of Physics ays, photoelectric effect, Compton roduction, intensity, X-ray imaging, rs nethods in medicine – Medical tracers, era, and PET

Week 25	Weekly learning outcomes	Student book links	Practical activity links
 Medical tracers and the gamma camera Positron emission tomography (PET) scanning Magnetic resonance Magnetic resonance imaging (MRI) scans Other techniques – endoscopes Other technologies – Doppler effect (qualitative) and blood flow 	 Students should be able to: Describe the use of medical tracers such as technetium-99m to diagnose the function of organs. Describe the main components of a gamma camera. Describe the principles of positron emission tomography (PET). Outline the principles of magnetic resonance, with reference to precession of nuclei, Larmor frequency, resonance and relaxation times. Describe the main components of a magnetic resonance imaging (MRI) scanner. Outline the use of MRI to obtain diagnostic information about internal organs. Describe the need for non-invasive techniques in diagnosis. Explain what is meant by the Doppler effect. Explain qualitatively how the Doppler effect can be used to determine the speed of blood. 	G485 Fields, Part 5.4.2 Diagnosis m	Work topic outlines ticles and Frontiers of Physics ethods in medicine – Magnetic /IRI, and the Doppler effect - Ultrasound

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Week 26	Weekly learning outcomes	Student book links	Practical activity links
 Ultrasound Piezoelectric effect Ultrasound scans Acoustic impedance Impedance matching 	 Students should be able to: Describe the properties of ultrasound. Describe the piezoelectric effect. Explain how ultrasound transducers emit and receive high-frequency sound. Describe the principles of ultrasound scanning. Describe the difference between A-scan and B-scan. Calculate the acoustic impedance using the equation Z = ρc. Calculate the fraction of reflected intensity using the equation I_I/I₀ = (Z₂-Z₁)²/(Z₂+Z₁)². Describe the importance of impedance matching. Explain why a gel is required for effective ultrasound imaging techniques. 	G485 Fields, Part 5.4.3 Ultrasound -	Work topic outlines cicles and Frontiers of Physics - Ultrasound, the Doppler effect, effect, ultrasound scanning, and edance

Week 27	Weekly learning outcomes	Student book links	Practical activity links
 Components in the universe Scale of the universe and astronomical measurements Stars – basic introduction Life cycle of stars – star formation Life cycle of stars – star death (small/average stars) Life cycle of stars – star death (large stars) Olbers' paradox 	 Students should be able to: Describe the principal contents of the universe – including stars, galaxies and radiation. Describe the solar system in terms of the Sun, planets, planetary satellites and comets. Describe the formation of a star such as our Sun from interstellar dust and gas. Describe the Sun's probable evolution into a red giant and white dwarf. Describe how a star much more massive than our Sun will evolve into a super red giant and then either into a neutron star or black hole. Define distances measured in astronomical units (AU), parsecs (pc) and light-years (ly). State the approximate magnitudes in metres of the parsec and light-year. State Olbers' paradox. Interpret Olbers' paradox to explain why it suggests that the model of an infinite, static universe is incorrect. 	G485 Fields, Part 5.5.1 Structure of universe, so	Work topic outlines ticles and Frontiers of Physics the universe – Components of the lar system, formation and evolution of uring distances, and Olbers' paradox

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Week 28	Weekly learning outcomes	Student book links	Practical activity links
 Week 28 1. Hubble's observations 2. Doppler effect – quantitative 3. Hubble's law 4. Cosmological principle 5. Cosmic background radiation 	 Weekly learning outcomes Students should be able to: Select and use the equation Δλ /λ = v/c. Describe and interpret Hubble's red shift observations. State and interpret Hubble's law. Convert the Hubble constant H₀ from its conventional units (km s⁻¹ Mpc⁻¹) to SI (s⁻¹). State the cosmological principle. Describe and explain the significance of the 3K microwave background radiation. 	Iinks Practical activity links • 2.5.5–6 • OCR Scheme of Work topic outlines G485 Fields, Particles and Frontiers of Physics 5.5.1 Structure of the universe – Hubble's law, the Hubble constant, cosmological principle, and microwave background radiation 5.5.2 The evolution of the universe – big bang model	

Week 29	Weekly learning outcomes	Student book links	Practical activity links
 The big bang theory Age of the universe Mean density Critical density and the fate of the universe 	 Students should be able to: Explain that the standard (hot big bang) model of the universe implies a finite age for the universe. Select and use the expression age of universe ≈ 1/H₀. Describe qualitatively the evolution of the universe 10⁻⁴³s after the big bang to the present. Explain that the universe may be open, flat or closed, depending on its density. Explain that the ultimate fate of the universe depends on its density. Define the term critical density. Select and use the expression for critical density of the universe p₀ = 3H₀²/8πG. Explain that it is currently believed that the density of the universe is close to, and possibly exactly equal to, the critical density needed for a flat cosmology. 	G485 Fields, Part 5.5.2 The evolution universe, ope	Work topic outlines ticles and Frontiers of Physics n of the universe – Evolution of the en universe, closed universe, flat d critical density